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Light Induced Molecular Changes ...

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Solutions Manual Chemistry: Matter and Change • Chapter 11  
209 Stoichiometry Stoichiometry CHAPTER 11 SOLUTIONS  
MANUAL Section 11.1 Defining Stoichiometry pages 368–372  
Practice Problems pages 371–372 1. Interpret the following  
balanced chemical equations in terms of particles, moles, and  
mass. Show that the law of conservation of mass is

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## **StoichiometryStoichiometry**

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In fact, multiple bonds of this type dominate the chemistry of the period 3 elements of groups 15 and 16. Multiple bonds to phosphorus or sulfur occur as a result of d-orbital interactions, as we discussed for the  $\text{SO}_4^{2-}$  ion in Section 4.5. In contrast, silicon in group 14 has little tendency to form discrete silicon-oxygen double bonds.

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