Chapter 11 Review Gases Section 3 Modern Chemistry Answers

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Chapter 11 Review Gases Section
SECTION 1 Date CHAPTER 11 REVIEW Gases Class SHORT ANSWER Answer the following questions in the space provided. b Pressure — orce For a constant force, when the surface area is tripled the surface area pressure is (a) doubled. as much. (c ripled. 7-0 (d) unchanged. Rank the following pressures in increasing order. (c) 76 torr (a) 50 kPa O, OOlctbv-x

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CHAPTER 11 REVIEW Gases SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of 1 mol. (c) multiplied by 22.4 L. (b) divided by the mass of 1 mol. (d) divided by 22.4 L. 2.

Chapter 11 Review Gases Section 1 Answers
CHAPTER 11 REVIEW . Gases . SHORT ANSWER . Answer the followin9 questions in the space provided. 1. c The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of I mol. (c) multiplied by 22.4 . L. (b) divided by the mass of 1 mol. (d) divided by 22.4 . L. 2. c For the expression V = n~T,

CHAPTER REVIEW Gases
462 Chapter 11 Gases Discovering the Relationships Between Properties If we want to explain why a weather balloon carrying instruments into the upper atmosphere expands as it rises, we need to consider changes in the properties of the gases (pressure, volume, temperature, or number of gas particles) inside and outside the balloon.

Chapter 11 Gases - An Introduction to Chemistry
CHAPTER 11 REVIEW. Molecular Composition of Gases. MIXED REVIEW. SHORT ANSWERAnswer the following questions in the space provided. 1. cThe average speed of a gas molecule is most directly related to the . (a)polarity of the molecule. (b)pressure of the gas. (c)temperature of the gas. (d)number of moles in the sample.

11 Molecular Composition of Gases - Madison Public Schools
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Properties Of Gases Section Review Chapter 11 Review Gases Section 1 Answers Section Goals and Introductions Section 111 Gases and Their Properties Goals To describe the particle nature of both real and ideal gases To describe the properties of gases that can be used to explain their characteristics: volume, number of particles, ...

[DOC] Properties Of Gases Section Review Answer
Chapter 11 Test Review. multiple choice (25) definition & applications of pressure (also atmospheric) SI unit of force. definition & use of a barometer. standard temperature & pressure (STP) Definition of Dalton’s law of partial pressures. Definitions & formulas for Boyle’s, Charles’, Gay-Lussac’s, and combined gas laws

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Modern Chemistry 97 Gases CHAPTER 11 REVIEW Gases SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. _____ The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of 1 mol. (c) multiplied by 22.4 L. (b) divided by the mass of 1 mol. (d) divided by 22.4 L. 2. _____ For the expression , P

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Chapter 11 181 Chapter 11 - Gases Review Skills 111 Gases and Their Properties Chapter 11 Map Chapter Checklist Read the Review Skills section If there is any skill mentioned that you have not yet neon gas mixed with argon gas (Objs 21 & 22) a If the total pressure of the mixture of gases is 130 kPa and the partial pressure of neon

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SECTION 3 Date CHAPTER 11 REVIEW Gases Class SHORT ANSWER Answer the following questions in the space provided c c The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of 1 mol (b) divided by the mass of 1 mol nRT (c) multiplied by 224 L (d) divided by 224 L For the expression V = (a) increasing P

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